



Supporting Information

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Phosphorus Enhanced Intermolecular Interactions of SnO₂ and Graphene as an Ultrastable Lithium Battery Anode

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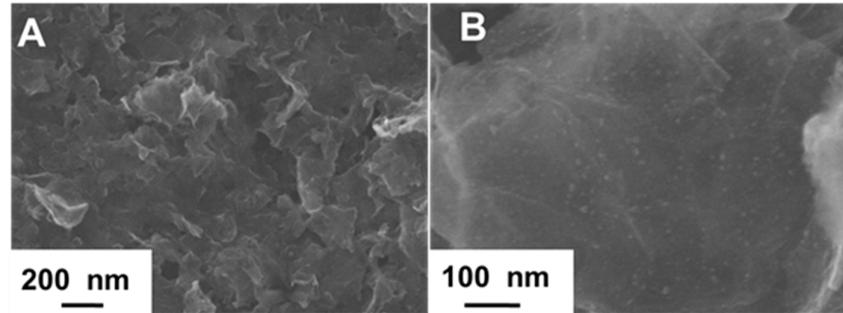


Figure S1. SEM images of SnO₂@P@GO.

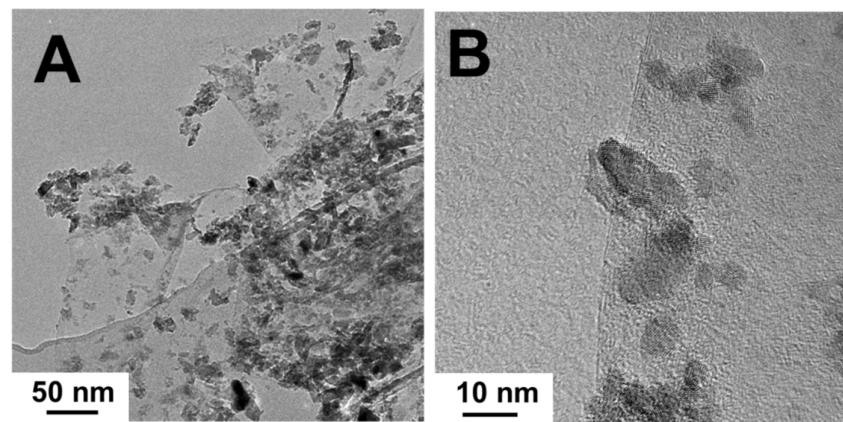


Figure S2. TEM images of SnO₂/GO

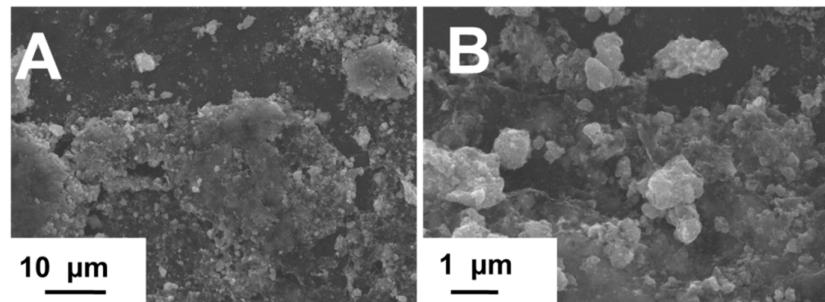


Figure S3. SEM images of SnO₂/P/GO

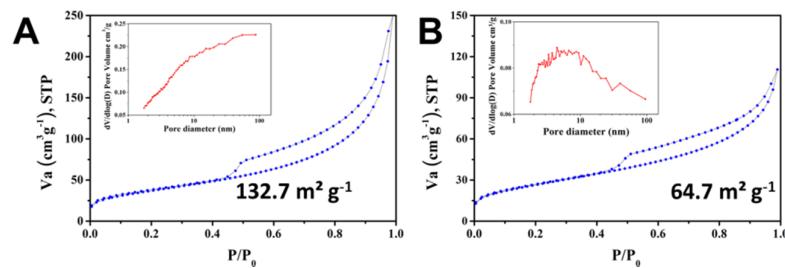


Figure S4. Nitrogen adsorption-desorption isotherms and pore size distributions (inset) of A) SnO₂@P@GO and B) SnO₂/P/GO.

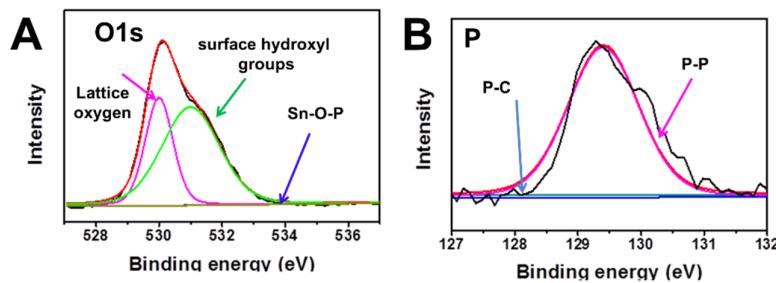


Figure S5. XPS spectrum of SnO₂/P/GO: A) O1s spectrum, B) P1p spectrum.

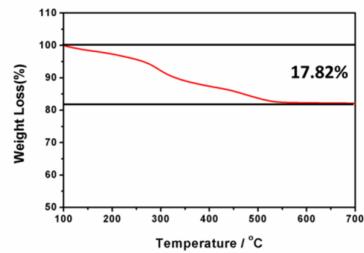


Figure S6. TG curves of SnO₂@P@GO in air.

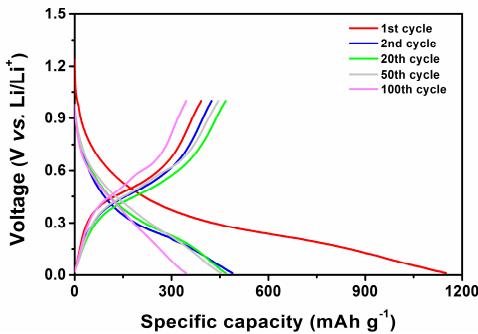


Figure S7. Charge-discharge curves of $\text{SnO}_2/\text{P}/\text{GO}$ at 100 mA g^{-1} .

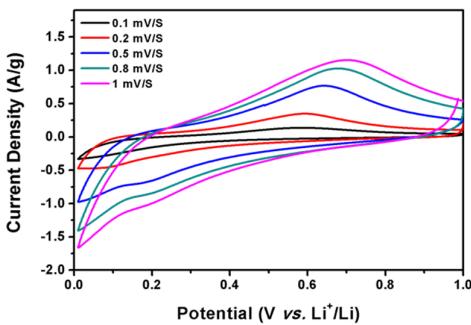


Figure S8. CV curves of $\text{SnO}_2@\text{P}@\text{GO}$ at different scan rate ranging from 0.1 to 1 mV s^{-1} .

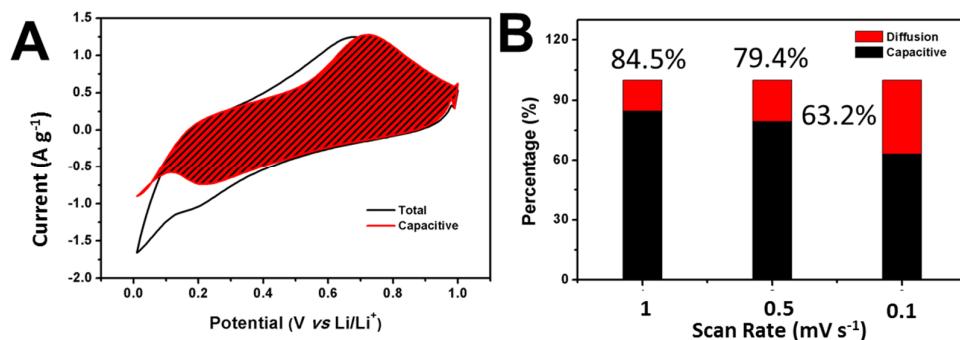


Figure S9. Kinetic analysis of $\text{SnO}_2@\text{P}@\text{GO}$: A) Separation of the capacitive and diffusion currents at a scan rate of 1 mV s^{-1} . B) Contribution ratio of the capacitive and diffusion-controlled charge at various scan rates.

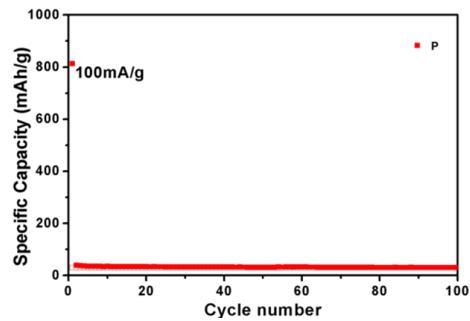


Figure S10. Cycling performance of bare P at 100 mA g⁻¹.

Table S1 ICP result

Mass Ratio	
Sn	5.4
P	1

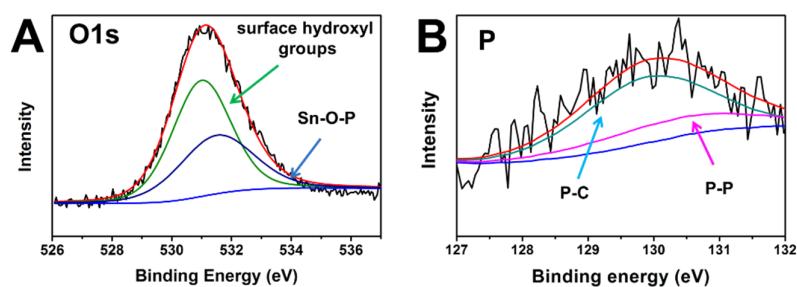


Figure S11. *Ex-situ* chemical composition analysis by XPS for A) O and B) P.