



Supporting Information

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Mesoporous $\text{Li}_3\text{VO}_4/\text{C}$ Submicron-Ellipsoids Supported on
Reduced Graphene Oxide as Practical Anode for High-Power
Lithium-Ion Batteries

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Keywords: lithium-ion battery; anode; mesoporous; high-power; Li_3VO_4

The SI file includes detailed Experimental Methods, 5 Tables and 9 Figures.

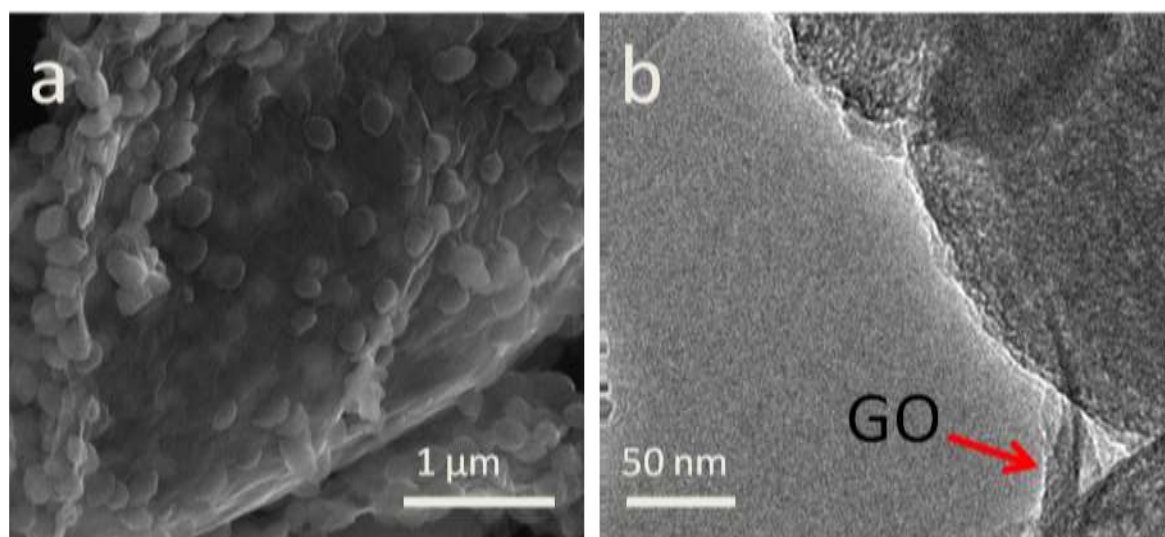


Figure S1. (a) SEM and (b) TEM images of LVO-EG/GO.

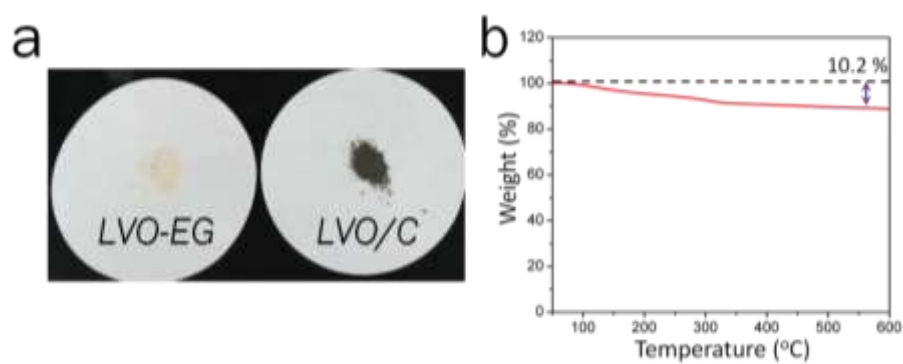


Figure S2. (a) Optical images of LVO-EG and LVO/C. (b) Thermogravimetric analysis (TGA) of LVO-EG in Ar atmosphere.

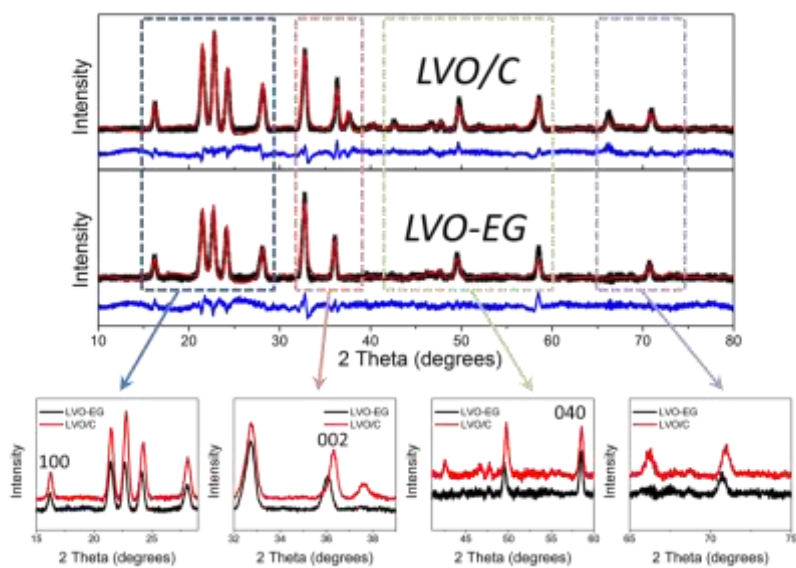


Figure S3. XRD patterns and the refinement XRD patterns of LVO/C and LVO-EG.

Table S1 Cell parameters of LVO-EG and LVO/C.

	a (Å)	b (Å)	c (Å)	cell volume (Å ³)
LVO-EG	6.309	5.452	4.972	171.020
LVO/C	6.310	5.446	4.940	169.759

Table S2 FT-IR band assignments of the four samples

LVO-EG	LVO/C	LVO-EG/GO	LVO/C/rGO	Band assignment	Reference
3428	3430	3420	3435	$\nu(\text{OH})$ water	34
		2949		$\nu(\text{CH})_{\text{as}}$	34
		1721		$\nu(\text{COO})$	35
1638		1622		$\delta(\text{OH})$	34
1496	1488	1478	1476	Amorphous carbon	34
	1425	1428	1426	Amorphous carbon	35
1404				(CH ₂) deformation	34
1314				$\delta(\text{CH}_2)$ in-plane	34
		1246		$\nu(\text{C-O})$ ether	34
		1088		$\nu(\text{CO})$ phenolic	34
1047		1046		$\nu(\text{C-O})$ alcoholic	34
842	831	841	828	$\nu(\text{V-O-V})$	33
472	464	476	465	$\nu(\text{V-O-V})$	33

ν : stretching; δ : bending.

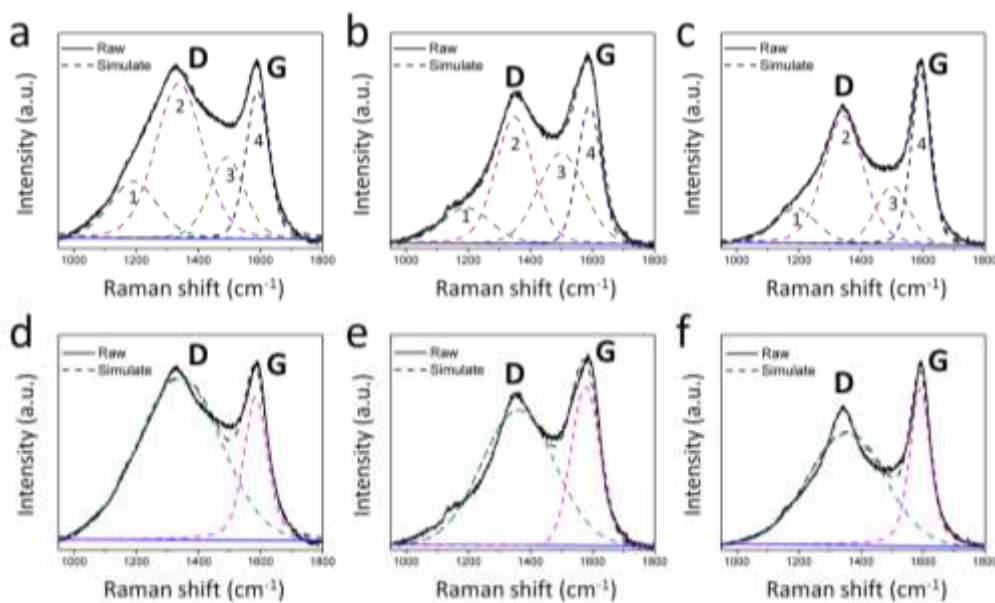


Figure S4. The Raman spectrum of LVO/C (a,d), LVO-EG/GO (b,e) and LVO/C/rGO (c,f).

Table S3 The I_D/I_G and d_{dom} values of LVO/C, LVO-EG/GO, and LVO/C/rGO.

	LVO/C	LVO-EG/GO	LVO/C/rGO
I_D/I_G	3.30	2.16	2.15
d_{dom} (nm)	11.676	17.839	17.922

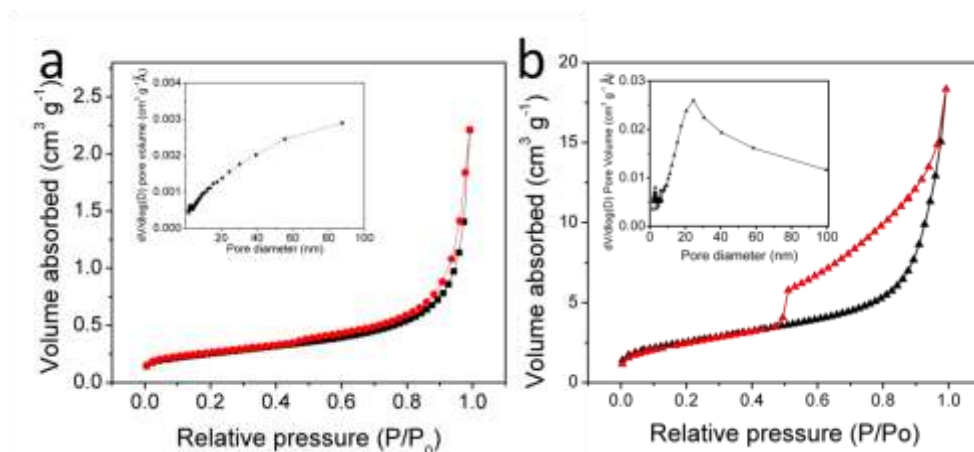


Figure S5. Nitrogen adsorption-desorption isotherms and pore size distribution curves of solid LVO particle (a) and LVO/C (b).

Table S4 The BET surface area and pore volume of the solid LVO, LVO/C, and LVO/C/rGO.

	solid LVO	LVO/C	LVO/C/rGO
BET surface area ($\text{m}^2 \text{g}^{-1}$)	0.89	9.0	12.5
Pore volume ($\text{cm}^3 \text{g}^{-1}$)	0.003	0.028	0.037

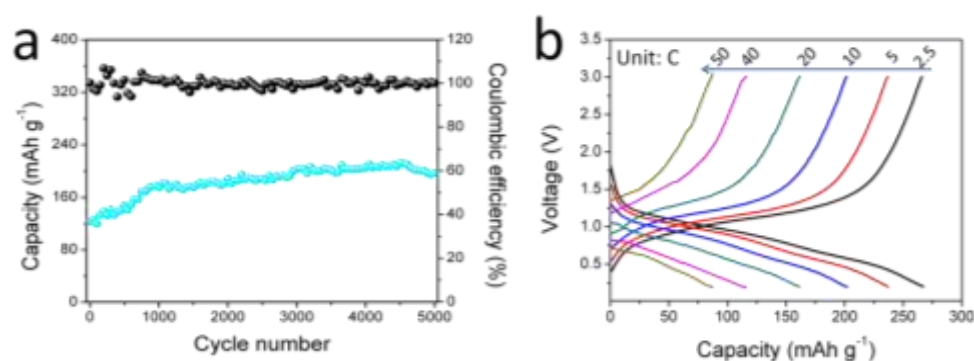


Figure S6. (a) Cycling performance of LVO/C anode at 10 C. (b) Discharge/charge curves of LVO/C anode at different current densities. (c) Rate performance of solid LVO/rGO electrode.

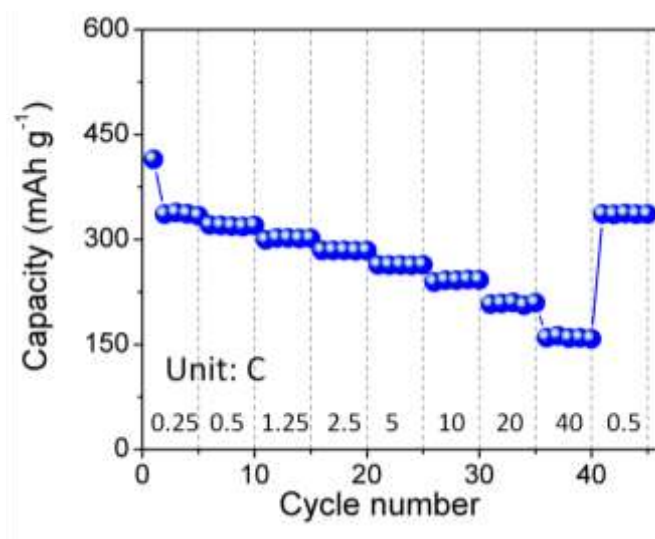


Figure S7. Rate performance of solid LVO/rGO electrode.

Table S5 The comparison of the electrochemical performances with precious works.

References	Anode	Cycling performance	Rate capability
Nanoscale, 2014, 6 11072	LVO/CNT	250 mAh g ⁻¹ after 2000 cycles at 5 C	240 mAh g ⁻¹ at 40 C
Chem. Commun. 2015, 51, 229	LVO@GNS	163 mAh g ⁻¹ after 5000 cycles at 5 C	133 mAh g ⁻¹ at 50 C
Nano Lett. 2013, 13, 4715	LVO/G	215 mAh g ⁻¹ after 500 cycles at 10 C	233 mAh g ⁻¹ at 20 C
Adv. Energy Mater. 2013, 3, 428	LVO	280 mAh g ⁻¹ after 25 cycles at 0.05 C	323 mAh g ⁻¹ at 0.2 C
Adv. Funct. Mater. 2015, 25, 3497	LVO/C	250 mAh g ⁻¹ after 2000 cycles at 10 C	106 mAh g ⁻¹ at 80 C
J. Electrochem. Chem. 2015, 745, 1	LVO/C	398 mAh g ⁻¹ after 80 cycles at 1C	370 mAh g ⁻¹ at 15 C
J. Power Sources 2013, 244, 557	LVO	185.55 mAh g ⁻¹ after 100 cycles at 1 C	249.14 mAh g ⁻¹ at 1 C
Nano Energy 2015, 12, 709	LVO/G	299 mAh g ⁻¹ after 1000 cycles at 10 C	88.4 mAh g ⁻¹ at 50 C
J. Power Sources 2015, 274, 345	LVO/C	394 mAh g ⁻¹ after 100 cycles at 1 C	100 mAh g ⁻¹ at 30 C
J. Power Sources 2014, 248, 122	LVO	396 mAh g ⁻¹ after 100 cycles at 1 C	216 mAh g ⁻¹ at 4 C
Our work	LVO/C/rGO	325 mAh g⁻¹ after 5000 cycles at 10 C	230 mAh g⁻¹ at 125 C

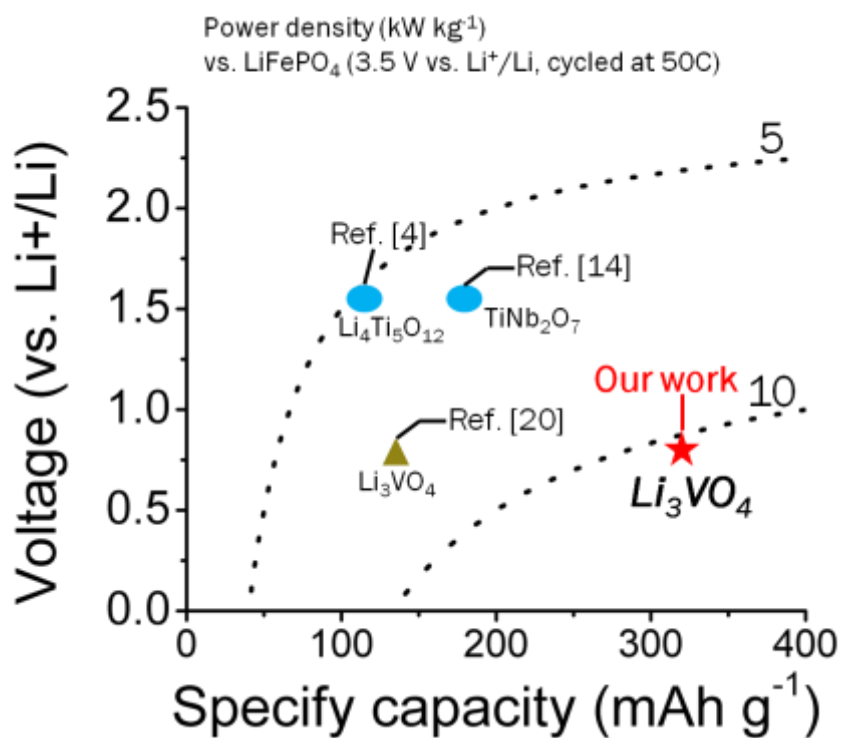


Figure S8. The energy density distribution of the three intercalation/deintercalation anodes, assuming LiFePO_4 (100 mAh g^{-1} at 50 C)^{ref. S1} as the cathode material.

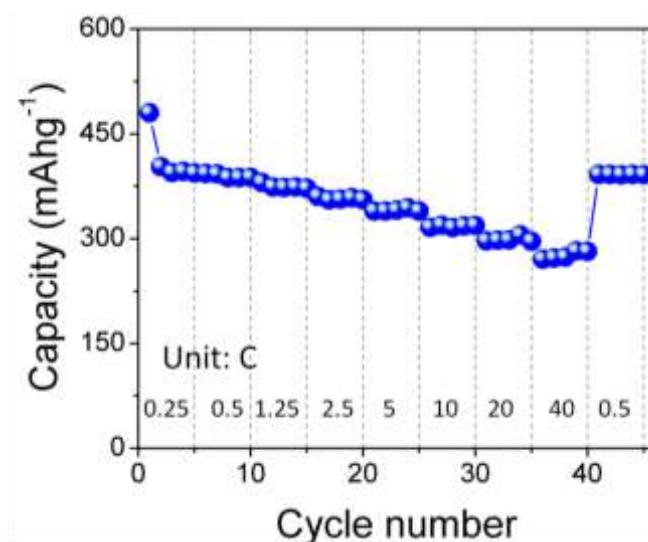


Figure S9. Rate performance of LVO/C/rGO with the composition of active material: acetylene black: binder = 92:3:5 in the composite anode.

References

- S1. B. Wang, W. A. Abdulla, D. Wang and X. S. Zhao, *Energy Environ. Sci.*, **2015**,8, 869-875