Electronic Supplementary Information

Enveloping SiO_x in N-Doped Carbon for Durable Lithium Storage via an Eco-

Friendly Solvent-Free Approach

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Fig. S1 Digital photos of raw materials before curing (a), SiO_x@NC precursor before calcination (b), and SiO_x@NC powder (c).



Fig. S2 Digital photos showing the scalable synthesis of $SiO_x@NC$.



Fig. S3 XRD patterns of $SiO_x@NC$ and bulk SiO_x .



Fig. S4 PSD curves of SiO_x@NC and SiO_x-BM.



Fig. S5 N₂ adsorption/desorption isotherms of $SiO_x@NC$ (a) and bulk SiO_x (b).



Fig. S6 High-resolution Si2p XPS spectrum of bulk SiO_x.



Fig. S7 High-resolution C1s (a), N1s (b), and O1s (c) XPS spectra of $SiO_x@NC$.



Fig. S8 Selected galvanostatic charge-discharge profiles of NC at 200 mA $g^{-1}(a)$, charge-discharge curves of NC at various current densities (b).



Fig. S9 Selected galvanostatic charge-discharge profiles of bulk SiO_x at 200 mA g⁻¹ (a), chargedischarge curves of bulk SiO_x at various current densities (b).



Fig. S10 Charge-discharge curves of $SiO_x@NC$ at various current densities.



Fig. S11 Long-term cycling performance of $SiO_x@NC$ with a lower carbon content at 500 mA g⁻¹.

Ref.	Reversible Capacity (mAh g ⁻¹)	Cycling Performance (mAh g ⁻¹)	Rate Capability (mAh g ⁻¹)	Electrochemical Window
This work	774 (200 mA g ⁻¹)	112 % (500 mA g ⁻¹ , 500 cycles)	345 (5 A g ⁻¹)	0.01 – 1.5 V
[S1]	570 (100 mA g ⁻¹)	≈ 102 % (100 mA g ⁻¹ , 100 cycles)	673 (800 mA g ⁻¹)	0.01 – 2.5 V
[S2]	906 (100 mA g ⁻¹)	≈ 80 % (100 mA g^{-1}, 350 cycles)	410 (800 mA g ⁻¹)	$0.0 - 3.0 \ V$
[83]	530 (500 mA g ⁻¹)	≈ 70 % (500 mA g $^{-1}$, 500 cycles)	231 (2 A g ⁻¹)	0.01 – 3.0 V
[S4]	1032 (100 mA g ⁻¹)	≈ 104 % (500 mA g ⁻¹ , 150 cycles)	309 (1 A g ⁻¹)	0.01 – 3.0 V
[85]	1107 (200 mA g ⁻¹)	≈ 133 % (1 A g ⁻¹ , 1000 cycles)	532 (2 A g ⁻¹)	0.01 – 3.0 V
[S6]	645 (65 mA g ⁻¹)	90 % (325 mA g ⁻¹ , 500 cycles)	549 (3.25 A g ⁻¹)	0.005 – 2.0 V
[S7]	965 (100 mA g ⁻¹)	91 % (500 mA g ⁻¹ , 400 cycles)	620 (600 mA g ⁻¹)	0.01 – 3.0 V
[S8]	1168 (100 mA g ⁻¹)	≈ 99 % (500 mA g $^{-1}$, 500 cycles)	725 (1 A g ⁻¹)	0.01 – 3.0 V
[89]	653 (120 mA g ⁻¹)	≈ 76 % (300 mA g $^{-1}$, 500 cycles)	582 (3 A g ⁻¹)	$0.005 - 2.0 \ V$
[S10]	765 (500 mA g ⁻¹)	79 % (200 mA g ⁻¹ , 200 cycles)	350 (5 A g ⁻¹)	0.01 – 2.0 V

Tab. S1 Lithium storage performances of various SiO_x -based anode materials.

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Fig. S12 The electrochemical impedance spectroscopy plots (a) and their results (b) of $SiO_x@NC$ and bulk SiO_x before cycling, the inset of (a) is equivalent circuit for fitting impedance plot.



Fig. S13 The electrochemical impedance spectroscopy plots (a) and their results (b) of $SiO_x@NC$ and bulk SiO_x after 100 cycles at 200 mA g⁻¹, the inset of (a) is equivalent circuit for fitting impedance plot.



Fig. S14 Top-view SEM images of bulk SiO_x before (a) and after (b) 100 cycles at 200 mA g^{-1} , top-view SEM images of SiO_x@NC before (c) and after (d) 100 cycles at 200 mA g^{-1} .



Fig. S15 HAADF-STEM images and EDS mappings of bulk SiO_x (a-d) and SiO_x @NC (e-h) after 100 cycles at 200 mA g⁻¹.



Fig. S16 Cross-sectional SEM images of bulk SiO_x -based electrode before (a) and after (b) 100 cycles at 200 mA g⁻¹, cross-sectional SEM images of SiO_x @NC-based electrode before (c) and after (d) 100 cycles at 200 mA g⁻¹.



Fig. S17 Selected galvanostatic charge-discharge profiles (a) and cycling performance (b) of LiFePO₄ at 0.2 C (1 C = 170 mA g⁻¹), charge-discharge curves (c) and rate performance (d) of LiFePO₄ at various current densities.