Supplementary Information

Novel Charging-Optimized Cathode for Fast and High-Capacity Zinc-Ion Battery

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Figure S1. SEM images of $K_2V_3O_8$ synthesized with different concentrations of V_2O_5 : (a) 0.8 mmol; (b) 1.2 mmol; (c) 1.8 mmol; (d) 2.4 mmol; (e) 3.6 mmol (with a fixed molar proportion, KOH : $V_2O_5 = 4 : 3$).



Figure S2. (a) The charge-discharge curve of $K_2V_3O_8$ for the first cycle, (b) the CV curves at scan rates of 0.1, 0.2, 0.3 and 0.4 mV s⁻¹ in 0.3 - 1.6 V and (c) the peak currents versus scan rates to determine the *b*-value of the anodic and cathodic peaks of $K_2V_3O_8$ cathode.



Figure S3. The charge-discharge curve of $Zn//K_2V_3O_8$ battery at different rates.





Figure S4. Comparison of rate performance with other representative Zn-storage cathode materials.



Figure S5. Waterfall plot of *in-situ* XRD ranging from 23° -35.5° (a), and waterfall plot of *in-situ* XRD ranging from 44.5°-50° (b).



Figure S6. Waterfall plot of electrochemical in-situ Raman spectra of $K_2V_3O_8$ cathode.



Figure S7. TGA results of electrodes at the original state (a), 1.6 V-charged-state (b) and 0.3 V-discharged-state (c).





Figure S8. XPS spectrum for all detected elements of the original K₂V₃O₈ cathode slice.