## Supporting Information

## A 3.0 V High Energy Density Symmetric Sodium-Ion Battery: Na<sub>4</sub>V<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>||Na<sub>3</sub>V<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>

Xuhui Yao,<sup>†</sup> Zixuan Zhu,<sup>†</sup> Qi Li,<sup>†</sup> Xuanpeng Wang,<sup>†</sup> Xiaoming Xu,<sup>†</sup> Jiashen Meng,<sup>†</sup> Wenhao Ren,<sup>\*,†</sup> Xinhe Zhang,<sup>§</sup> Yunhui Huang,<sup>‡</sup> and Liqiang Mai<sup>\*†</sup>

<sup>†</sup>State Key Laboratory of Advanced Technology for Materials Synthesis and Processing, International School of Materials Science and Engineering, Wuhan University of Technology, Wuhan 430070, P. R. China

\*Email: mlq518@whut.edu.cn; 13871303279@whut.edu.cn

<sup>‡</sup>State Key Laboratory of Materials Processing and Die and Mould Technology, School of Materials Science and Engineering, Huazhong University of Science and Technology, Wuhan 430074, P. R. China

<sup>§</sup>Dong Guan McNair New Power Co., Ltd, Guangdong, Dong Guan 523000, P. R. China



Figure S1. Crystal structure of  $Na_3V_2(PO_4)_3$  (NVP).



Figure S2. Energy dispersive spectrometry elemental mapping of NVP/C.



Figure S3. Raman spectrum of NVP/C.



Figure S4. Nitrogen adsorption-desorption isotherms (inset is the corresponding pore

size distribution) of NVP/C.



Figure S5. The anode materials obtained via discharge-charge process with coins cell.



Figure S6. The morphological of (a) the NVP/C materials and (b) the anode material

with pre-intercalation.



**Figure S7.** The GITT curves of  $Na_4V_2(PO_4)_3$  in the third cycle.



Figure S8. EIS results of Na<sub>4</sub>V<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> electrode, inset is the calculation of Warburg

factor.



Figure S9. Typical voltage profiles of NVP/C as (a) cathode and (b) anode in

different current densities.



Figure S10. CV curves of the full cell in the initial three cycles.



Figure S11. The lighted LED bulb driven by the full cell.



Figure S12. Rate performance of the symmetric full cell.