

Supporting Information for
Facile and scalable synthesis of $\text{Zn}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$
microflowers as a high-performance anode for lithium-ion
batteries

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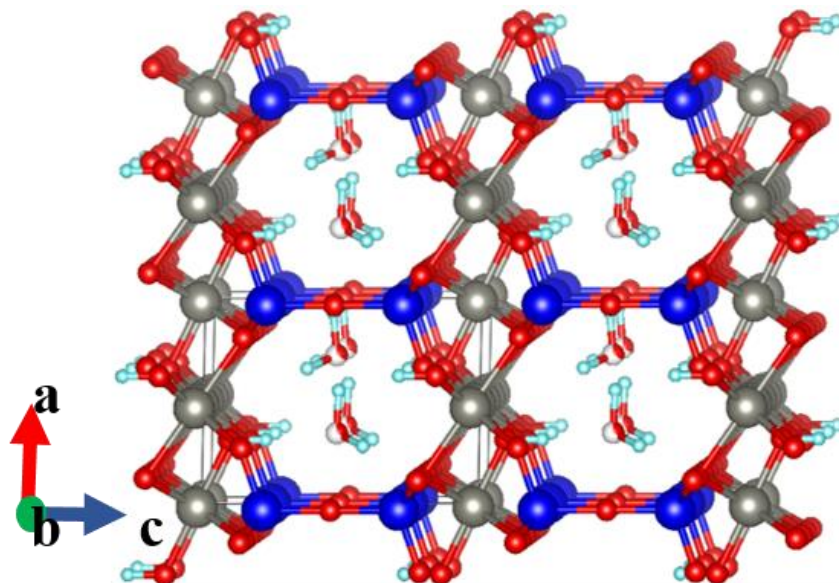


Figure S1. Crystal structure of $\text{Zn}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$ (a,b,c refers to the a,b,c axis of crystal cell; grey: Zn; dark blue: V; red: O; light blue: H; molecule in the layers: H_2O ;))

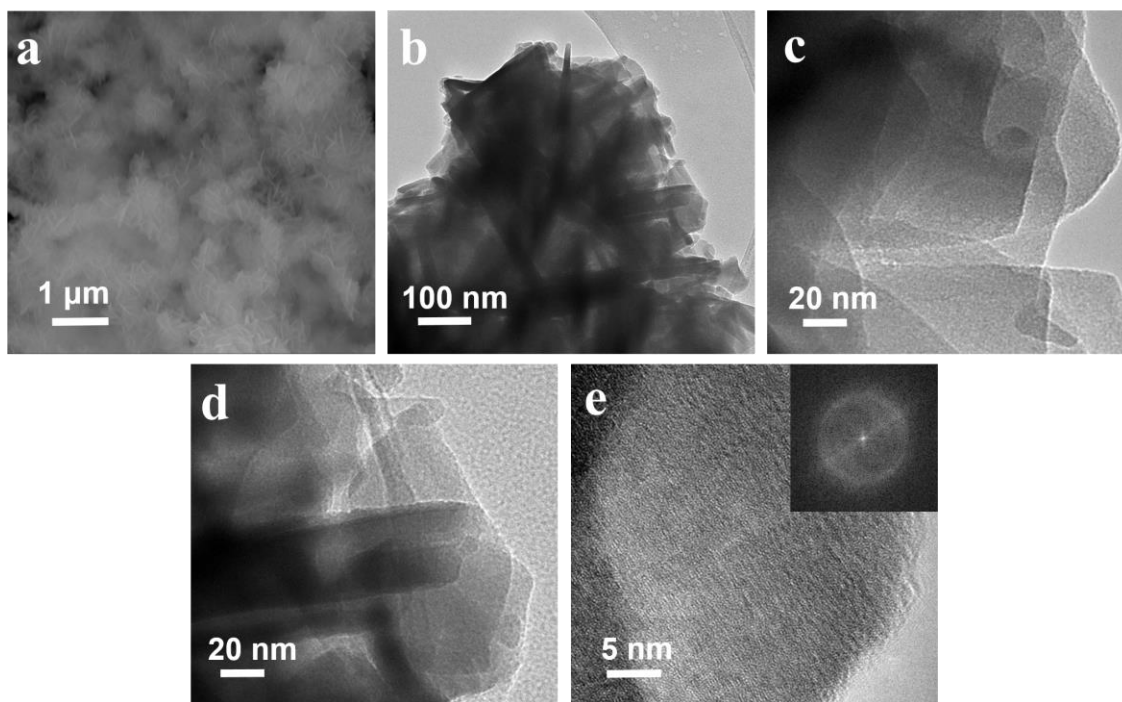


Figure S2. (a) SEM image of ZnVO-2. (b-e) TEM, HRTEM images, and FFT pattern of ZnVO-2.

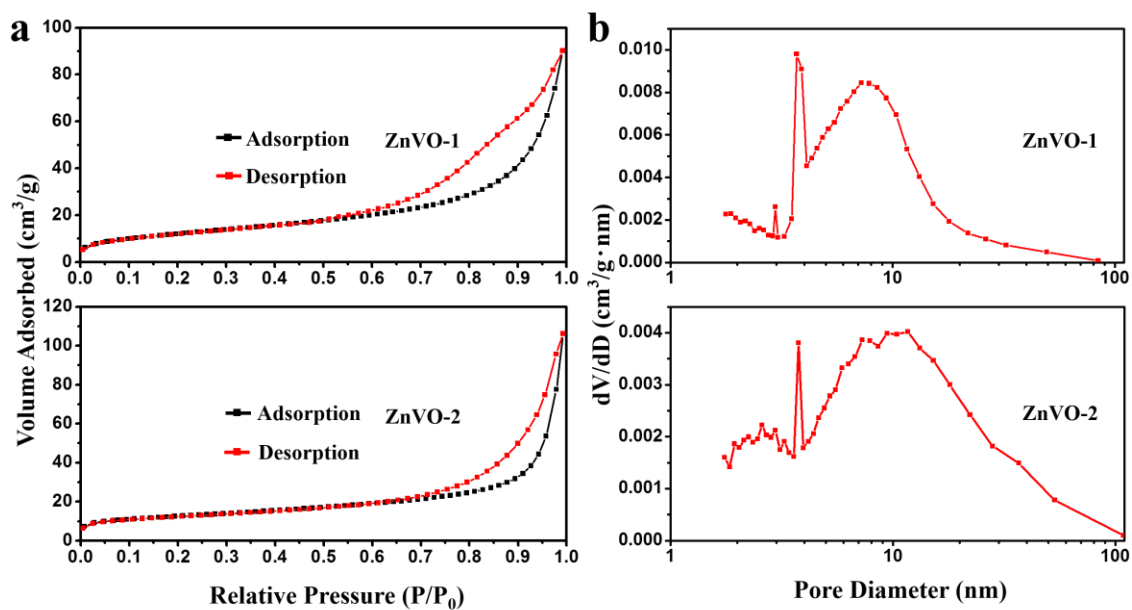


Figure S3. (a) The N₂ adsorption-desorption isotherms of ZnVO-1 and ZnVO-2. (b) The corresponding BJH pore size distribution curves of ZnVO-1 and ZnVO-2.

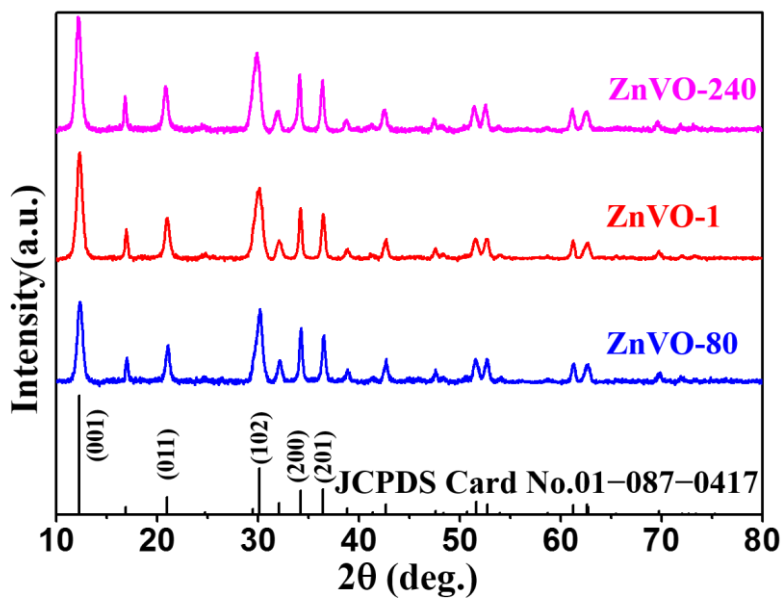


Figure S4. XRD patterns of ZnVO-80, ZnVO-1, and ZnVO-240.

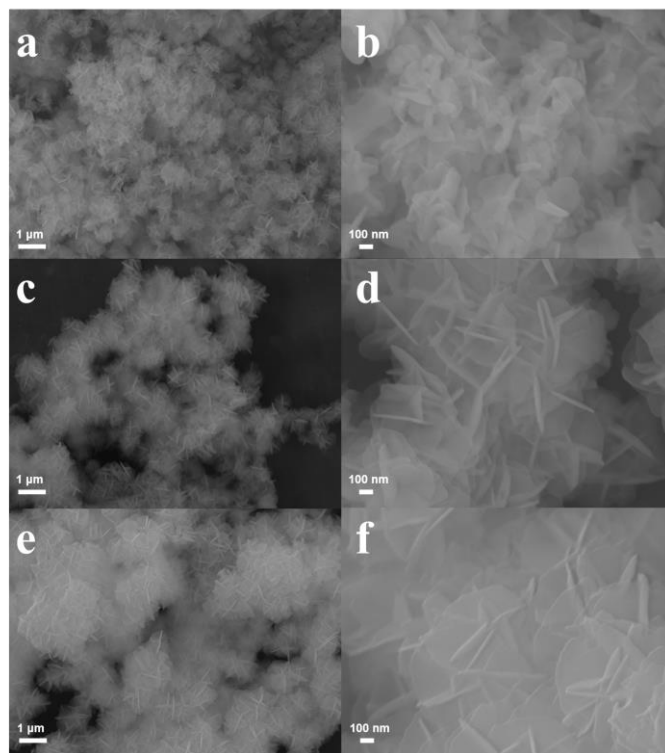


Figure S5. SEM images of (a,b) ZnVO-80, (c,d) ZnVO-1, and (e,f) ZnVO-240.

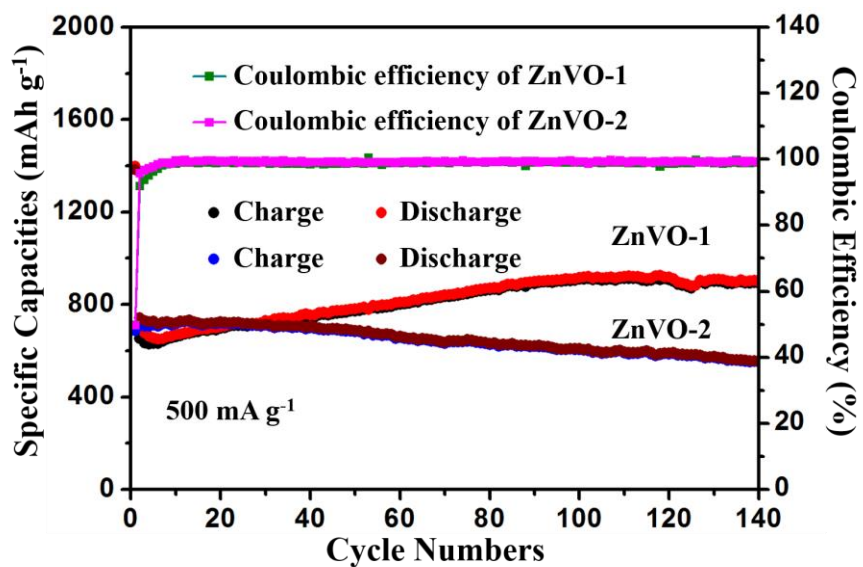


Figure S6. The electrochemical performance of the ZnVO-1 and ZnVO-2 at current density of 500 mA g^{-1} .

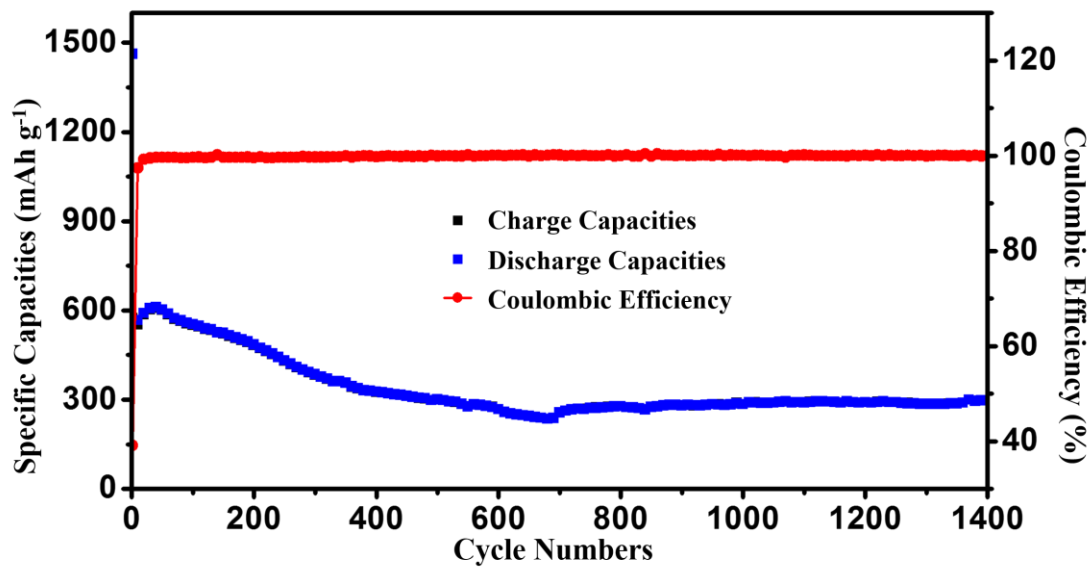


Figure S7. Long cycling performance and corresponding Coulombic efficiency of ZnVO-1 at 5 A g^{-1} .

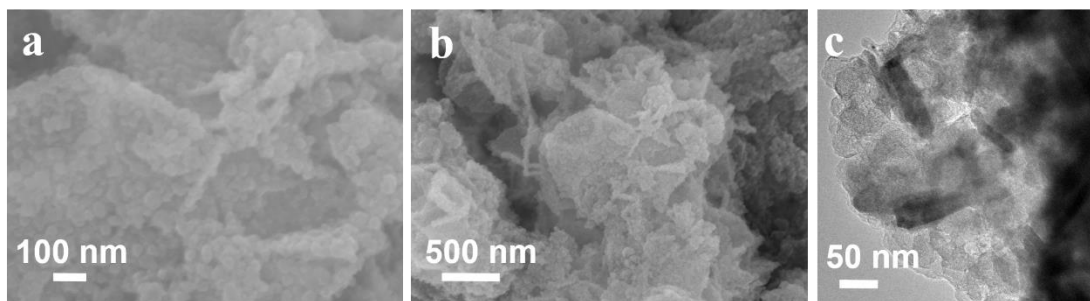


Figure S8. (a, b) SEM images and (c) TEM image of the charged-state ZnVO-1 electrode after 100 cycles.

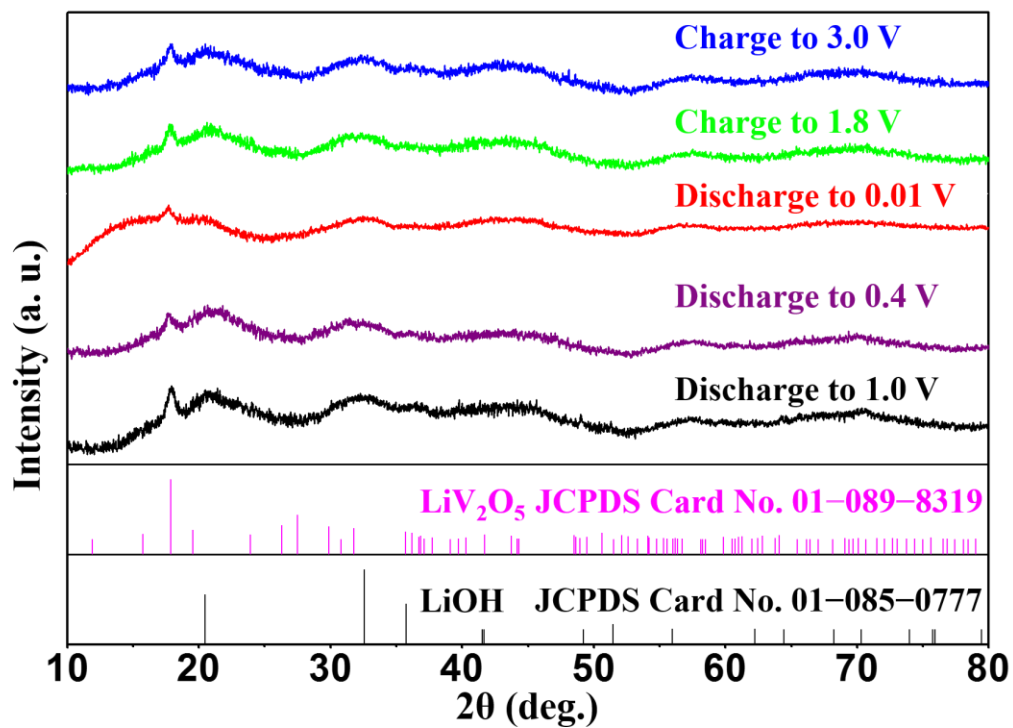


Figure S9. Ex situ XRD patterns of ZnVO-1 electrode at different states in the 30th cycle.

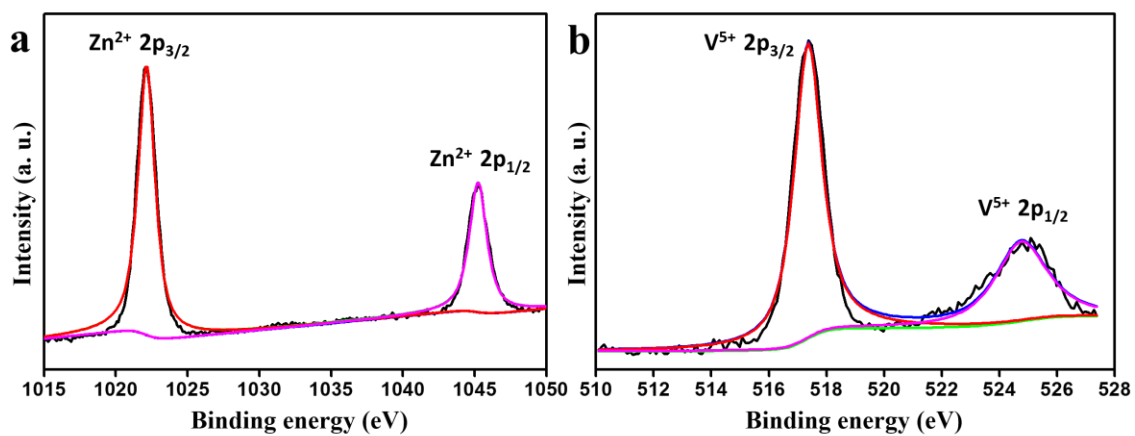


Figure S10. High resolution (a)Zn 2p XPS spectra and (b)V 2p XPS spectrum of fresh ZnVO-1 electrode.

Table S1. Comparison of specific capacities between our work and pervious works

Electrode	Reversible capacity (mA h g⁻¹)	Current density (mA g⁻¹)	Cycle numbers	Mass ratio	Voltage range (V vs Li⁺/Li)
ZnVO-1	1287	200	140	7:2:1	0.01-3
	931	500	500	7:2:1	0.01-3
Zn₃V₂O₇(OH)₂·2H₂O graphene network¹	930	200	200	7:2:1	0.01-3
Zn₃V₂O₇(OH)₂·2H₂O nanobelts²	750	20	20	7:2:1	0.02-2.5
Zn₃V₂O₇(OH)₂·2H₂O 3D microspheres³	619	20	20	8:1:1	0.02-2.5
Zn₃V₂O₈ nanocages⁴	1400	100	80	6:3:1	0.005-3
Zn₃V₂O₈ hexagon nanosheets⁵	1103	200	150	6:3:1	0.01-3
Ultralong monoclinic ZnV₂O₆ nanowires⁶	973	100	10	4.5:5:0. 5	0.025-3
Zn₃V₃O₈/C microspheres⁷	912	400	150	7:2:1	0.01-3
ZnV₂O₄ microspheres⁸	501	100	100	8:1:1	0.02-3
Zn₃V₂O₈ 3D microspheres³	492	20	20	8:1:1	0.02-2.5
Zn₃V₂O₈ nanoplatelets⁹	270	100	40	8:1:1	0.01-3

The mass ratio in the table refers to the mass ratio of active materials, conducting materials and binder

Reference

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